

## 7 - The Psychrometric Chart

The Psychrometric Chart is a plot of the thermodynamic variables of moist air. Each variable depends only on the Humidity Ratio HR (or Absolute Humidity or Specific Humidity) and on the air temperature  $t_{\text{DRY}}$  (the dry bulb temperature or air temperature). The plot is shown in the Excel spreadsheet [7-psy-chart.xls](#)

Starting with equation (5) section (4),

$$\text{HR} = \frac{m_{\text{VAPOR}}}{m_{\text{DRY-AIR}}} = \left( \frac{M_{\text{VAPOR}}}{M_{\text{DRY-AIR}}} \right) \left( \frac{p_{\text{VAPOR}}}{p_{\text{ATM}} - p_{\text{VAPOR}}} \right) \quad (1)$$

As noted,  $p_{\text{VAPOR}}$  can be expressed as a function of two variables only: HR and the atmospheric pressure  $p_{\text{ATM}}$ :

$$p_{\text{VAPOR}} = \frac{p_{\text{ATM}}}{1 + \left( \frac{M_{\text{VAPOR}}}{M_{\text{DRY-AIR}}} \right) \left( \frac{1}{\text{HR}} \right)} \quad (2)$$

Observe that in this equation  $p_{\text{VAPOR}}$  does not depend on temperatures.

$M_{\text{VAPOR}}$  = molecular mass of water vapor = 18 grams

$M_{\text{DRY-AIR}}$  = molecular mass of dry air = 29 grams

$M_{\text{VAPOR}} / M_{\text{DRY-AIR}} = 0.62069$

### 1. Plotting the wet bulb temperature $t_{\text{WET}}$

From equation (10) section (4):

$$p_{\text{VAPOR}} = p_{\text{SAT}}(t_{\text{WET}}) - k (t_{\text{DRY}} - t_{\text{WET}}) p_{\text{ATM}}$$

For each value of  $t_{\text{WET}} = 0, 5, \dots, 30^{\circ}\text{C}$ , compute  $p_{\text{VAPOR}}(t_{\text{DRY}})$  and then  $\text{HR}(t_{\text{DRY}})$  from eq. (1) above. Plot the curve  $\text{HR}(t_{\text{DRY}})$  for each value of  $t_{\text{WET}}$ .

$$k = 0.000662 / ^{\circ}\text{C}$$

### 2. Plotting the Relative Humidity RH

Start from equation (11) section (4):

$$\text{RH \%} = 100 \frac{p_{\text{VAPOR}}}{p_{\text{SAT}}(t_{\text{DRY}})}$$

and using equation (2) above, HR can be computed as:

$$\boxed{\text{HR} = \frac{\left( \frac{M_{\text{VAPOR}}}{M_{\text{DRY-AIR}}} \right)}{\left( \frac{100 p_{\text{ATM}}}{\text{RH}\% p_{\text{SAT}}(t_{\text{DRY}})} - 1 \right)}} \quad (3)$$

For each value of RH%, plot the curve HR( $t_{\text{DRY}}$ )

$$M_{\text{VAPOR}} / M_{\text{DRY-AIR}} = 0.62069$$

### 3. Plotting the Specific Volume SV

Specific Volume =  $\text{m}^3$  of air / kg of dry air

Starting from eqs. (3) and (4) of section (4):

$$\text{HR} = \frac{m_{\text{VAPOR}}}{m_{\text{DRY-AIR}}} = \left( \frac{p_{\text{VAPOR}} V_{\text{IN}} M_{\text{VAPOR}}}{R T_{\text{DRY}}} \right) \left( \frac{1}{m_{\text{DRY-AIR}}} \right) = \left( \frac{V_{\text{IN}}}{m_{\text{DRY-AIR}}} \right) \left( \frac{p_{\text{vapor}} M_{\text{VAPOR}}}{R T_{\text{DRY}}} \right)$$

$$\text{SV} = \text{specific volume} = \frac{V_{\text{IN}}}{m_{\text{DRY-AIR}}}$$

$$\text{HR} = \text{SV} \left( \frac{p_{\text{vapor}}}{273.15 + t_{\text{DRY}}} \right) \left( \frac{M_{\text{VAPOR}}}{R} \right)$$

Using equation (2) above and solving for HR:

$$\boxed{\text{HR} = \frac{\text{SV} M_{\text{VAPOR}} p_{\text{ATM}}}{R (273.15 + t_{\text{DRY}})} - \frac{M_{\text{VAPOR}}}{M_{\text{DRY-AIR}}}}$$

R = universal constant of gases = 0.06237 ( $\text{m}^3 \text{mmHg} / ^\circ\text{C}$ )

$$M_{\text{VAPOR}} / M_{\text{DRY-AIR}} = 0.62069$$

$$M_{\text{VAPOR}} = 18 \text{ grams}$$

### 4. Plotting the dew point temperature ( $t_{\text{DEW}}$ )

Given the dew point temperature, the relative humidity is determined, according to equation (3) section (5):

$$\text{RH}\% = 100 \frac{p_{\text{S}}(t_{\text{DEW}})}{p_{\text{S}}(t_{\text{AMB}})}$$

Inserting this equation in equation (3) above, with  $t_{\text{AMB}} = t_{\text{DRY}}$ :

$$HR = \frac{\left( \frac{M_{\text{VAPOR}}}{M_{\text{DRY-AIR}}} \right)}{\left( \frac{p_{\text{ATM}}}{p_{\text{SAT}}(t_{\text{DEW}})} - 1 \right)}$$

which represents horizontal lines for each value of  $t_{\text{DEW}}$ .

$$M_{\text{VAPOR}} / M_{\text{DRY-AIR}} = 0.62069$$

## 5. Plotting the Enthalpy h

The Enthalpy used in the Psychrometric Chart is the total amount of heat energy, per kilogram of dry air, needed to heat the moist air from 0°C (liquid water) to  $t_{\text{DRY}}$ . This amount is (according to Thermodynamics):

To vaporize liquid water: 2445 (kjoules / kg of water)

To heat water vapor from 0° to  $t_{\text{DRY}}$ : 2.006  $t_{\text{DRY}}$  (kjoules / kg of water)

Energy needed per kg of dry-air:

$$(2445 + 2.006 t_{\text{DRY}}) \text{ HR} \quad (\text{kjoules} / \text{kg of dry air})$$

To heat the dry air from 0° to  $t_{\text{DRY}}$ :

$$1.022 t_{\text{DRY}} \text{ (kjoules} / \text{kg of dry air)}$$

Specific Enthalpy h:

$$h = 1.022 t_{\text{DRY}} + (2445 + 2.006 t_{\text{DRY}}) \text{ HR}$$

To plot the curves for each value of **h**, solve above equation to **HR**:

$$HR = \frac{h - 1.022 t_{\text{DRY}}}{2445 + 2.006 t_{\text{DRY}}}$$

## 6. An example of a commercial psychrometric chart

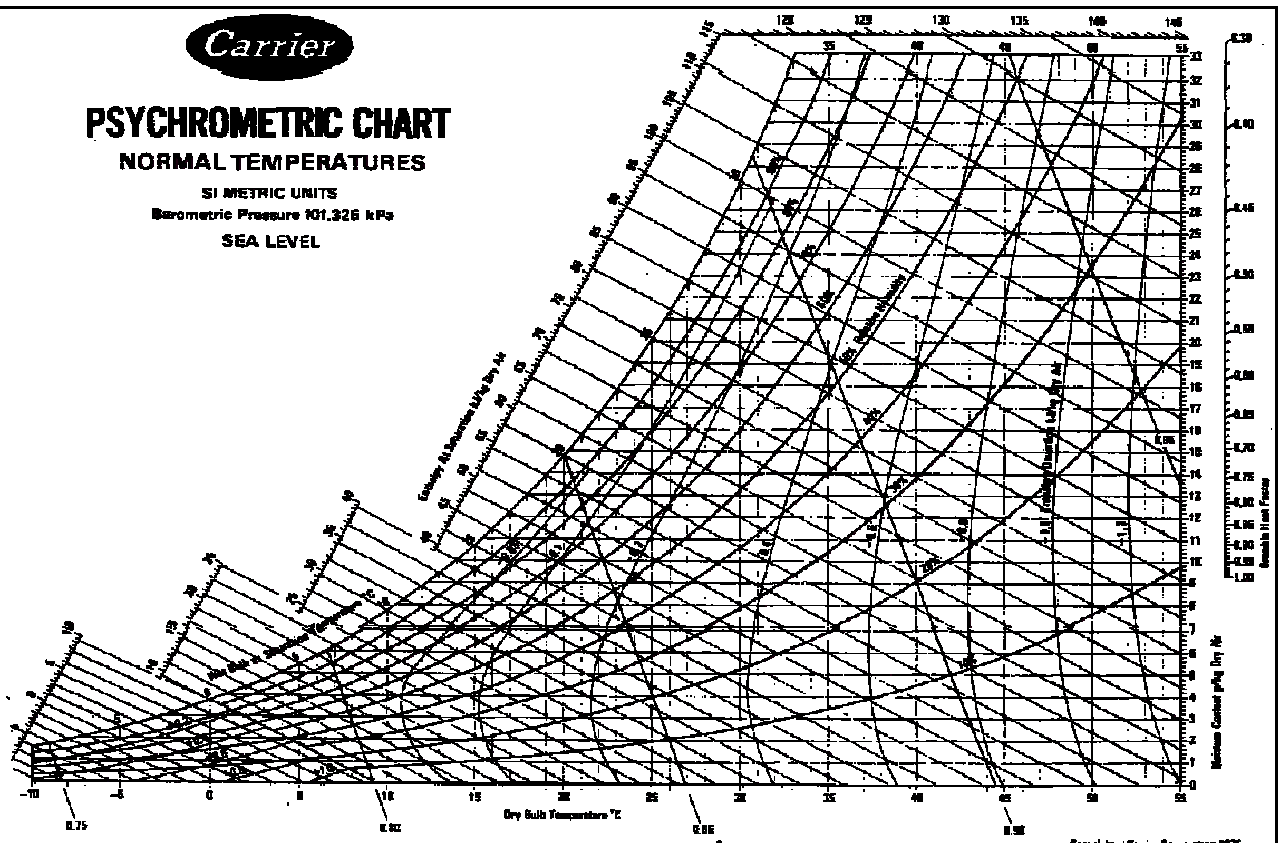
Here is an example of a very useful chart, published on paper by the well known company Carrier. This section (7) has shown how it may be created using the Excel spreadsheet "Psy-Chart-XLS.xls", with the advantage that it may be easily customized according to user needs.



# PSYCHROMETRIC CHART

NORMAL TEMPERATURES

SI METRIC UNITS  
Barometric Pressure 101.325 kPa  
SEA LEVEL



Below 0°C Properties and Enthalpy Deviation Lines/Ratio For Ice

Volume m³/kg Dry Air

Rev. 8.81

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